

Chemquest 53 Answers

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 53. ChemQuest 17. Name: ____ Date: ____ Hour: ____ Information: Separating Charges. Examine Figure 1 below where there are three pairs of metal spheres that have different amounts of charge on them. The spheres in diagram C are closer than the others because they have the strongest attraction.

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HW Keys - Roosevelt High School AP Chemistry 2017-18
 This video is designed to help you complete the molarity chemquest found in Unit 11.

Molarity
 118 ChemQuest 39 Name: ____ Date: ____ Hour: ____ Information: Molarity Concentration is a term that describes the amount of solute that is dissolved in a solution. Concentrated solutions contain a lot of dissolved solute, but dilute solutions contain only a little.

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 An element exists as 4 different isotopes. 4.35% have a mass of 49.9461, 83.79% have a mass of 51.9405, 9.50% have a mass of 52.9407 amu, and 2.36% have a mass of 53.9389 amu. a. What is the average atomic mass for this element? b. What is the identity of this element?

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 15. Given your answer to question 14 and the reaction you wrote in question 13, calculate the molarity of the nitrate ions (NO₃⁻) in the solution. Title: Molarity ChemQuest WS.pages Created Date:

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 (Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of CO₂ are produced, so the number of moles of CO₂ that can be produced when 4 moles of propane combusts = $4(3) = 12$.