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Use the mass m of the ring and the atomic mass M of gold to calculate the number of gold atoms. Each atom has 79 protons and an equal number of electrons. SET UP: $23 \text{ NA } 6.02 \times 10^{23} \text{ atoms/mol}$. A proton has charge $+e$.

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Substitute 150 V for V , 10 nC for q_1 , and 20 nC for q_2 in the equation and solve for r . Therefore, the electric potential to move 20 nC charge from A to B is same as the electric potential to move 10 nC charge from A to B and the value is V . Comment (1) olivert mbah.

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i dont have the answers, but when i was stumped by one of the questions, i would just try to google it. Try it, it might work for you... I usually found the answers. also, try and post the question in the math section of yahoo answers, there are a bunch of braniacs that love physics

Does anyone have the rest of the answers to Mastering Physics?

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